

Macropolyhedral boron-containing cluster chemistry. Assessment of the possibilities of thermolytic mixed-cluster fusion, and of the use of synchrotron X-radiation for the examination of small single crystals of metallaboranes. Isolation and structure of eighteen-vertex [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀]*

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Thermolysis of [(CO)(PMe₃)₂HrB₈H₁₂] with molten B₁₀H₁₄ resulted in small yields of eighteen-vertex [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀] for which a small crystal was examined by single-crystal diffraction analysis, using synchrotron X-radiation, in order to determine the molecular structure.

Macropolyhedral boron-containing cluster compounds in which the multicentre bonding characteristic of boranes extends contiguously over two or more subclusters that are fused together is currently an expanding area of new chemistry. In any novel area of chemistry exploratory work is necessary to reveal the extent of structure and reactivity available. Reaction types are unknown, and in initial exploratory work yields are often small. Nevertheless it is essential that the chemistry be pursued in order to establish the structural parameters available in a new area so that focused rational and targeted chemistry may thence be developed to define the new field. In this context there are often severe difficulties in structural characterisation if only very small quantities of new products are available, and suitable techniques have to be developed. Here we report a new eighteen-vertex metallaborane [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀], obtained in very small yield from a mixed-cluster fusion reaction between [(CO)(PMe₃)₂HrB₈H₁₂] and B₁₀H₁₄. Only one very small single crystal was ultimately obtained, and so synchrotron radiation in a new experimental facility was used to obtain sufficiently intense diffraction data for the precise determination of the molecular structure.

Macropolyhedral borane clusters derive from the fusion of smaller single-cluster species. Previously they have been generally synthesised from thermolytic auto-fusion of single compounds. Mixed fusions should provide a larger variety. Here we report results from a mixed cluster fusion between *arachno* nine-vertex [4-(CO)-4,4-(PMe₃)₂-4-H-4-IrB₈H₁₂] and *nido* ten-vertex B₁₀H₁₄. A set of *arachno* nine-vertex compounds^{1–3} constitutes the system so far known to be most prone to macropolyhedral generation by auto-fusion. Of these compounds, we have noted that auto-fusion and other reactions of [(PMe₂Ph)₂PtB₈H₁₂] and [(CO)(PMe₃)₂HrB₈H₁₂] show indications of reaction with their solvent media.^{3,6} Hence we are investigating the thermolysis of [(CO)(PMe₃)₂HrB₈H₁₂] in molten B₁₀H₁₄ as solvent, and report a preliminary result here.

The cluster compounds [(CO)(PMe₃)₂HrB₈H₁₂] (134 mg, 280 μmol)⁷ and B₁₀H₁₄ (500 mg, 4.1 mmol) were finely ground together and heated under dry dinitrogen at 134 °C (oil bath).

Excess B₁₀H₁₄ was removed by vacuum sublimation (water bath at 80 °C), and preparative TLC (silica gel G, CH₂Cl₂-hexane mixtures) showed many orange and yellow bands. Repeated TLC separations ultimately gave an orange band (*R_F* 0.7 using 50 : 50 CH₂Cl₂-hexane), which upon further separation by HPLC (silica, Lichrosorb Si60 7 μm; 260 × 16 mm column; rapid initial gradient 25% CH₂Cl₂ in hexane to 100% CH₂Cl₂, flow rate 4 cm³ min⁻¹) gave one major component *R_T* 32 min (*ca.* 500 μg), a compound that we currently formulate as a {B₂₆Ir₂} cluster species, and which we hope to be able to report on when we complete work upon it. The several minor HPLC components were each systematically screened for microcrystal formation by hexane diffusion into concentrated CDCl₃ solutions at +4 °C and one of these, with *R_T* 30 min, gave a crop of six very small orange crystals of [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀] [*ca.* 20 μg (estimate)]. One of these was isolated at room temperature for single-crystal diffraction work, but in doing so redissolution of the others occurred, so far irreversibly, into the solvent mixture thus precluding NMR assessment of the pure product.

The crystal was of size 0.25 × 0.10 × 0.08 mm. In our experience, crystals of metallaboranes of this type with dimensions *ca.* 0.1–0.5 mm often give relatively weak diffracted intensities with conventional equipment and require extended exposure times even with a CCD area detector. The crystal was therefore examined on a newly constructed single-crystal diffraction station at CCLRC Daresbury Laboratory, with wiggler-generated synchrotron radiation of wavelength 0.6956 Å (calibrated against samples of known cell parameters, monochromated and focused by a bent silicon crystal)⁸ and a specially adapted Siemens SMART CCD diffractometer.[†] Individual

[†] All measurements were made on a Siemens SMART CCD area-detector diffractometer at 160 K with narrow ω-rotation frames using synchrotron radiation (λ = 0.6956 Å; SRS station 9.8, Daresbury Laboratory). The structure was solved by heavy-atom methods using SHELXS 86⁹ and refined by full-matrix least squares (against all the unique *F*² data) using SHELXL 93.¹⁰ Non-hydrogen atoms were refined with anisotropic displacement parameters. Methyl hydrogen atoms were constrained to idealised positions (C–H 0.98 Å) whereas hydrogen atoms associated with the cluster were located in Fourier-difference syntheses and freely refined. The complex crystallised as a racemic twin; the twin component parameter¹¹ refined to 0.493(5). Crystal data for [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀], C₇H₃₈B₁₇IrOP₂, *M* = 576.28, crystal dimensions 0.25 × 0.10 × 0.08 mm, orthorhombic, space group *P*2₁2₁2₁, *a* = 10.5204(11), *b* = 14.6809(15), *c* = 15.863(2) Å, *Z* = 4, *U* = 2450.0(4) Å³, *D_c* = 1.56 g cm⁻³; 21 372 reflections were collected to θ = 26.00; 4704 unique reflections (*R_{int}* = 0.0395) were used in calculations after Lorentz, polarisation and absorption corrections (based on repeated and symmetry-equivalent reflections; μ = 5.579 mm⁻¹, transmission factors 0.374–0.491). Final *wR*² = {Σ[*w*(*F_o*² – *F_c*²)²]/Σ(*F_o*²)²}]^{1/2} = 0.040, conventional *R* = 0.018 for *F* values of 4591 reflections with *F_o*² > 2σ(*F_o*²). Weighting scheme *w* = 1/[σ²(*F_o*²) + 0.0201*P*²]

* 7-Carbonyl-7,7-bis(trimethylphosphine)-*nido*-decaborano-(5',6':5,6)-*nido*-7-iridadecaborane.

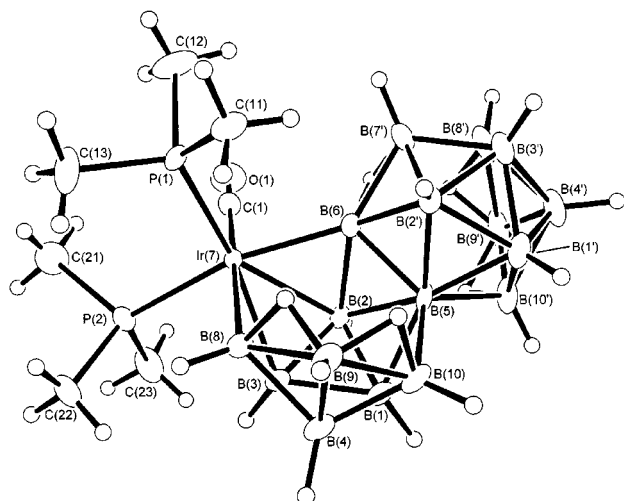
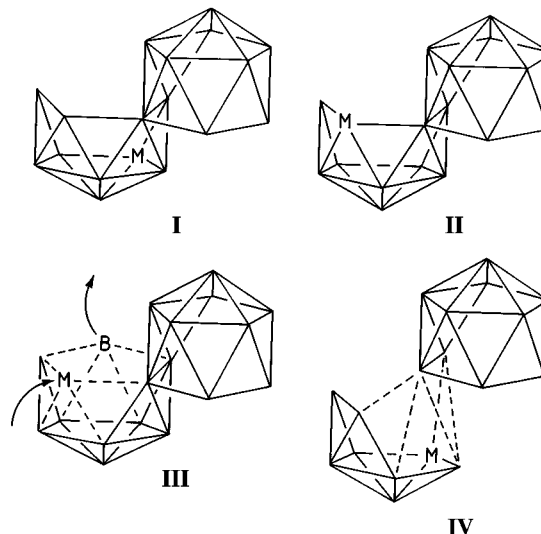


Fig. 1 An ORTEP-type drawing¹² of the molecular structure of [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀]. Some salient interatomic distances (Å) are: B(5)–B(6) 1.779(6), B(5)–B(10) 2.032(6), B(9)–B(10) 1.776(6), B(8)–B(9) 1.818(6), B(5)–B(10') 1.981(6), B(8')–B(9') 1.782(9), B(9')–B(10') 1.770(8), B(7')–B(8') 1.979(7) and B(7')–B(6) 1.805(5) [the other interboron distances being in the range 1.712(6)–1.846(6)], and from Ir(7) to B(2) 2.253(4), B(3) 2.235(4), B(6) 2.133(4), B(8) 2.295(4), C(1) 1.891(4), P(1) 2.3329(10) and to P(2) 2.3791(9). Angles (°) at Ir(7) between vectors are as follows: to C(1) and P(1) 95.33(12), C(1) and P(2) 85.27(11), P(1) and P(2) 100.53(4), B(6) and P(1) 97.47(11), and B(8) and P(1) 89.02(11). The longer interboron distances B(5)–B(10), B(5)–B(10') and B(7')–B(8') characterise *nido* ten-vertex structures. The short iridium–boron distance Ir(7)–B(6) has precedent in short interatomic distances in other ten-vertex *nido* (6,7)-type positions that lack hydrogen bridges¹³

exposures of 3 s gave strong diffraction effects, so that a complete set of data was collected in 1.25 h of X-ray exposure time (3 h elapsed time); further symmetry-equivalent sets of data were also measured. The crystal was cooled to 160 K and there was no evidence of significant decomposition, the observed steady decay in intensities being accounted for by the monitored synchrotron beam current decay. The final residual indices of $R=0.018$ and $wR2=0.040$ from the structure refinement are the lowest achieved so far from this experimental station and are excellent by the standards of conventional equipment. Even a much smaller crystal would probably have yielded perfectly satisfactory results.

The basic cluster structure of [7-(CO)-7,7-(PMe₃)₂-syn-7-IrB₁₇H₂₀] (Fig. 1) consists of *nido* {B₁₀} plus *nido* {IrB₉} ten-vertex subclusters fused with a common two-boron edge. The position of fusion generates a new eighteen-vertex metalla-octadecaborane of *syn-nido*: *nido* ten-vertex: ten-vertex macropolyhedral configuration (schematic I). The {Ir(CO)(PMe₃)₂} unit, formally octahedral iridium(III), contributes three electrons to the cluster bonding scheme, and thus generates a cluster analogous to the [syn-B₁₈H₂₁][−] monoanion. The *syn*-{7-IrB₁₇} configuration contrasts to the only other reported {MB₁₇} type, the {*syn*-10-MB₁₇} configuration (schematic II) seen in {RhB₁₇} and in {CoB₁₇} cluster formation from *anti*-B₁₈H₂₂ systems.¹⁴ These latter 10-metalla-octadecaborane structures II can be visualised as forming relatively simply *via* a boron-vertex pseudo-replacement from *anti*-B₁₈H₂₂-based residues (schematic III). However, a mechanism to generate the



7-hetero configuration I is not feasible by this or related pathways from either *syn* or *anti* B₁₈H₂₂-type residues without much fundamental cluster rearrangement. Rather, in the mixed-fusion reaction system, it may well occur *via* a loss of a boron vertex during the course of the reaction to generate an *arachno*-type of {MB₇} species which then adds to the *nido*-B₁₀H₁₄ substrate (schematic IV). There is precedent for the generation of seven-boron systems from eight-boron and nine-vertex systems.¹⁵ That the iridium atom is next to the *nido*-decaboranyl subcluster in the product suggests fusion *via* electrophilic attack by the metal centre in the 5,6 region of the B₁₀H₁₄ substrate.

In any event, it is clear that this mixed-fusion method can viably generate new macropolyhedral species not otherwise obtainable, albeit in very small yields so far. We currently explore for other mixed fusions, and also hope to report soon on other low-yield macropolyhedral products from this preliminary siting experimentation in the [(CO)(PMe₃)₂HIrB₈H₁₂]-B₁₀H₁₄ system, such as the {Ir₂B₂₆} species mentioned above. The excellent structural results show that the application of new synchrotron single-crystal diffraction facilities provides an effective means of characterising materials obtainable only as very small crystals, a situation by no means uncommon in metallaborane chemistry. We are currently exploring the practical limits of the technique.

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where $P = (F_o^2 + 2F_c^2)/3$, goodness of fit = 1.035 for all F^2 values and 341 parameters. Maximum and minimum residual electron density +0.90 and −0.68 e Å³ respectively. Atomic coordinates, thermal parameters, and bond lengths and angles have been deposited at the Cambridge Crystallographic Data Centre (CCDC). See Instructions for Authors, *J. Chem. Soc., Dalton Trans.*, 1997, Issue 1. Any request to the CCDC for this material should quote the full literature citation and the reference number 186/544.

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